

Article

Aurantioside L, a New Tetramic Acid Glycoside with Anti-Leishmanial Activity Isolated from the Marine Sponge *Siliquariaspongia japonica*

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Abstract: A new tetramic acid glycoside, aurantioside L (**1**), was isolated from the sponge *Siliquariaspongia japonica* collected at Tsushima Is., Nagasaki Prefecture, Japan. The structure of aurantioside L (**1**) composed of a tetramic acid bearing a chlorinated polyene system and a trisaccharide part was elucidated using spectral analysis. Aurantioside L (**1**) showed anti-parasitic activity against *L. amazonensis* with an IC₅₀ value of 0.74 μM.

Keywords: aurantiosides; *Siliquariaspongia japonica*; marine sponge; nuclear magnetic resonance; mass spectrometry; anti-leishmanial activity; marine natural products



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1. Introduction

Leishmaniasis are vector-borne parasitic diseases caused by several different species of *Leishmania* [1]. It is estimated that there are 12 million patients suffering from leishmaniasis with around 1 million new cases annually (www.who.int (accessed on 13 July 2023)). Visceral leishmaniasis, caused by *L. donovani* in Asia and Africa and *L. infantum* in the Mediterranean Basin, Middle East, Central Asia, South America, and Central America, is characterized by clinical symptoms such as fever, anemia, splenomegaly, hepatomegaly, and weight loss and is fatal unless treated appropriately [2]. Cutaneous leishmaniasis is a skin manifestation that sometimes heals naturally over 3–18 months, but the ulcer can lead to serious permanent scarring, disfigurement, and stigmatization [3]. Mucocutaneous leishmaniasis leads to the partial or total destruction of the mucous membranes of the nose, mouth, and throat, and the disease can be found mostly in South America (www.who.int (accessed on 13 July 2023)). Because no effective vaccines to prevent the disease for humans are commercially available yet, the control of the disease mostly relies on chemotherapy. These leishmaniasis are treated by, for example, pentavalent antimonials, liposomal amphotericin B, and miltefosine. However, drug side effects, high costs, potential teratogenicity, and the emergence of drug-resistant strains pose a serious potential threat to endemic countries where leishmaniasis are prevalent [4,5].

Consequently, there have been efforts to discover new candidate compounds for chemotherapeutic use against leishmaniasis. Although target-based screening is becoming

a more popular way for drug discovery, the phenotypic screening of natural compounds is still a vital choice [6]. In fact, marine bioproducts are attractive sources of anti-parasitic agents for various diseases including malaria and leishmaniasis [7]. Gracilioethers A-C were isolated from the marine sponge *Agelas gracilis* as anti-protozoan natural compounds, which had anti-malarial activity [8]. A xenicane diterpenoid, cristaxenicin A, was found in the deep-sea gorgonian *Acanthoprimnoa cristata*, showing anti-leishmanial and anti-trypanosomal activities (against *Leishmania amazonensis* and *Trypanosoma congolense*, respectively) [9].

To discover potential drug leads against leishmaniasis, we focused on marine invertebrates, whose extracts are a rich source of various bioactive compounds [10]. Marine organism extracts (1565 samples) have been tested against the recombinant *L. amazonensis* doped with a green fluorescent protein (*La/egfp*). In this screening approach, a marine sponge *Siliquariaspongia japonica* extract showed strong anti-parasitic activity against *La/egfp*. From the Lithistida order of sponges, to which *S. japonica* belongs, a wide variety of compounds that are thought to be produced by the symbiotic bacteria [11] have been isolated. The marine sponge *Siliquariaspongia* sp. has been also reported several times as the source of unique and bioactive metabolites, for example, aurantosides D-E [12], rubrosides A-H [13], motualevic acids A-F [14], mirabamides A-D [15], celebesides A-C, and theopapuamides B-D [16]. Based on the result of the screening and the abundant discoveries reported so far, we considered the marine sponge *S. japonica* to be a suitable candidate for searching for substances with anti-leishmanial activity. The bioassay-guided fractionation of this sponge extract provided a new anti-leishmanial tetramic acid glycoside, aurantoside L (1) (Figure 1). In this paper, the isolation, structure elucidation, and biological activities of this compound are discussed.

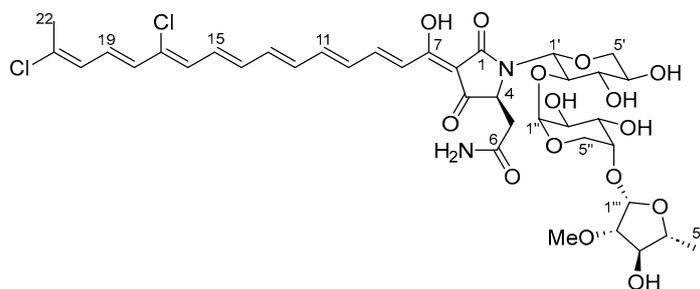


Figure 1. Structure of aurantoside L (1).

2. Results

The frozen specimen of *S. japonica* (170 g, wet weight) was extracted with MeOH and CHCl₃/CH₃OH (1:1) repeatedly. The extracts were combined, and the concentrated extract was partitioned between H₂O and CHCl₃. The water-soluble layer was further extracted with *n*-C₄H₉OH, and the *n*-C₄H₉OH layer was combined with the former CHCl₃ layer. The combined organic layer was fractionated using the Kupchan procedure [17] to yield *n*-hexane, CHCl₃, and 60% CH₃OH layers. The anti-leishmanial aqueous CH₃OH layer was subjected to octadecylsilyl (ODS) column flash chromatography to give six fractions (fr. A-F). Among them, active fr. E, eluting with 100% CH₃OH, was subjected to reversed-phase high-performance liquid chromatography (RP-HPLC) to afford compound 1 as the active substance (12.5 mg, 7.4 × 10⁻³% yield based on the wet weight).

Compound 1 was obtained as a red amorphous solid. The electrospray ionization mass spectrum (ESIMS) (positive mode) showed clusters of ions at *m/z* 865, 867, and 869 [M + Na]⁺ in the ratio of 9:6:1, indicating the presence of two chlorine atoms in 1 since the natural abundance of chlorine atoms is 75% with integer mass 35 and 25% with 37. The molecular formula was established as C₃₈H₄₈Cl₂N₂O₁₅ by high-resolution electrospray ionization mass spectrum (HRESIMS) (positive mode) (*m/z* 865.2310 [M + Na]⁺, calcd for C₃₈H₄₈³⁵Cl₂N₂O₁₅Na, 865.2324, Δ -1.6 mmu).

The ^1H NMR spectrum (in CD_3OD , at 297 K, 400 MHz) showed 12 downfield shifted signals $\{\delta_{\text{H}}$ (6.39, d, $J = 11.5$ Hz), (6.44, d, $J = 14.4$ Hz), (6.55, d, $J = 11.6$ Hz), (6.56, m), (6.59, m), (6.61, dd, $J = 14.5, 11.5$ Hz), (6.72, dd, $J = 16.1, 11.6$ Hz), (6.75, dd, $J = 14.4, 11.5$ Hz), (6.86, dd, $J = 14.2, 11.6$ Hz), (6.89, dd, $J = 14.5, 11.6$ Hz), (7.22, d, $J = 15.1$ Hz), and (7.60, dd, $J = 15.1, 11.5$ Hz)}; 17 signals in the range of δ_{H} 3.21–5.09 (typical for saccharides) $\{\delta_{\text{H}}$ (3.21, dd, $J = 11.0, 10.8$ Hz), (3.48, dd, $J = 8.8, 8.8$ Hz), (3.59, dd, $J = 12.9, 2.3$ Hz), (3.62, m), (3.67, dd, $J = 8.1, 4.4$ Hz), (3.74, dd, $J = 12.9, 0.9$ Hz), (3.76, qd, $J = 7.1, 6.4$), (3.79, dd, $J = 9.5, 2.5$ Hz), (3.80, dd, $J = 9.5, 2.9$ Hz), (3.88, dd, $J = 10.8, 4.8$ Hz), (3.90, dd, $J = 8.1, 7.1$ Hz), (3.91, m), (4.33, br), (4.52, br), (4.52, br), (5.04, br), and (5.09, d, $J = 4.4$ Hz)}; 1 vinyl methyl (δ_{H} 2.26, brs); 1 methoxy group (δ_{H} 3.35, s); and 1 doublet methyl group (δ_{H} 1.32, d, $J = 6.2$ Hz) (Supplementary Information).

The interpretation of the COSY spectrum revealed the two spin systems of H-8 to H-12 $\{\delta_{\text{H}}$ (7.22, d, $J = 15.1$ Hz)/(7.60, dd, $J = 15.1, 11.5$ Hz)/(6.61, dd, $J = 14.5, 11.5$ Hz)/(6.89, dd, $J = 14.5, 11.6$ Hz)} and H-18 to H-20 $\{\delta_{\text{H}}$ (6.44, d, $J = 14.4$ Hz)/(6.75, dd, $J = 14.4, 11.5$ Hz)/(6.39, d, $J = 11.5$ Hz)}. The HMQC and HMBC spectra established the polyene substructure **a** (C-8 to C-22), in which quaternary carbons C-17 and C-21 were chlorinated based on the ^{13}C chemical shift values (δ_{C} 134.7 and δ_{C} 136.4, respectively) [18]. The geometry of the double bonds in substructure **a**, except for Δ^{16} , was determined as *E* based on the proton–proton coupling constants ($^3J_{8,9} = 15.1$ Hz, $^3J_{10,11} = 14.5$ Hz, $^3J_{12,13} = 16.1$ Hz, $^3J_{14,15} = 14.2$ Hz, and $^3J_{18,19} = 14.4$ Hz), whereas that at Δ^{16} was deduced as *Z* because of the chlorine substitution at C-17. The chemical shift of C-22 (δ_{C} 21.8) and the NOE cross peak between H-19/H-22 indicated that the geometry at Δ^{20} is *E* [19,20] (Figure 2).

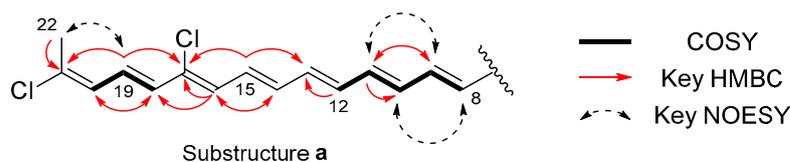


Figure 2. Substructure **a** deduced from COSY and key HMBC and NOESY correlations.

COSY cross peaks in the range of δ_{H} 3.21–5.09 (in CD_3OD , at 297 K, 400 MHz) showed four spin systems, H-3'/H-4'/H-5' $\{\delta_{\text{H}}$ (3.48, dd, $J = 8.8, 8.8$ Hz)/(3.62, m)/(3.21, dd, $J = 11.0, 10.8$ Hz)/(3.88, dd, $J = 10.8, 4.8$ Hz)}, H-1''/H-2'' $\{\delta_{\text{H}}$ (5.04, br)/(3.80, dd, $J = 9.5, 2.9$ Hz)}, H-3''/H-4''/H-5'' $\{\delta_{\text{H}}$ (3.79, dd, $J = 9.5, 2.9$ Hz)/(3.91, m)/(3.59, dd, $J = 12.9, 2.3$ Hz)/(3.74, dd, $J = 12.9, 0.9$ Hz)}, and H-1'''/H-2'''/H-3'''/H-4'''/Me-5 $\{\delta_{\text{H}}$ (5.09, d, $J = 4.4$ Hz)/(3.67, dd, $J = 8.1, 4.4$ Hz)/(3.90, dd, $J = 8.1, 7.1$ Hz)/(3.76, qd, $J = 7.1, 6.4$ Hz)/(1.32, d, $J = 6.4$ Hz)}, suggesting the existence of three sugar units (sugar-I, -II and -III). HMQC and HMBC spectra (in CD_3OD , at 297 K) indicated two of these sugars are a pyranose (sugar-II) and a 5-deoxypentofuranose (sugar-III). Sugar-II was determined to be an arabinopyranose, in which H-2'', H-3'', and H-5'' a were axial, and H-4'' and H-5'' b were equatorial based on the coupling constants ($^3J_{2'',3''} = 9.5$ Hz, $^3J_{3'',4''} = 2.9$ Hz, $^3J_{4'',5''a} = 2.3$ Hz, and $^3J_{4'',5''b} = 0.9$ Hz).

A methoxy group was located at C-2''' in sugar-III by the HMBC cross peaks of H-2'''/OCH₃ (δ_{H} 3.67, dd, $J = 8.1, 4.4$ Hz/ δ_{C} 58.5) and OCH₃/C-2''' (δ_{H} 3.35, s/ δ_{C} 87.4). The NOESY cross peaks among H-1'''/OCH₃ (δ_{H} 5.09, d, $J = 4.4$ / δ_{H} 3.35, s), OCH₃/H-3''' (δ_{H} 3.35, s/ δ_{H} 3.90, dd, $J = 8.1, 7.1$ Hz), and H-3'''/H-5''' (δ_{H} 3.90, dd, $J = 8.1, 7.1$ Hz/ δ_{H} 1.32, d, $J = 6.4$ Hz) revealed that sugar-III was a 5-deoxy-2-*O*-methylpentofuranose.

H-1'' and H-2'' signals in sugar-I were broadened in the ^1H NMR spectrum at 297 K, but the distinct HMQC cross peaks among H-1'/C-1' and H-2'/C-2' (δ_{H} 4.52, br/ δ_{C} 81.4 and δ_{H} 4.52, br/ δ_{C} 86.3) were observed at a higher temperature (in CD_3OD , at 320 K, 400 MHz). The assignment of H-1' and H-2' was not possible because of their overlapping signals; however, HMBC cross peaks between H-3'/C-2' (δ_{H} 3.48, dd, $J = 8.8, 8.8$ Hz/ δ_{C} 81.4) and H-1'/C-2' (δ_{H} 5.04, br/ δ_{C} 81.4) confirmed the assignment of C-2' at this position. Along with a NOESY cross peak to H-1'/H-5' (δ_{H} 5.04, br/ δ_{H} 3.88, dd, $J = 10.8, 4.8$ Hz), coupling constants among these proton signals suggested that sugar-I was a xylopyranose. An HMBC cross peak of H-1'''/C-4'' (δ_{H} 5.09, d, $J = 4.4$ Hz/ δ_{C} 79.7) indicated a sequential

connection of sugars-I/II/III through α (1 \rightarrow 2) and α,β (1 \rightarrow 4), respectively (substructure **c**) (Figure 3).

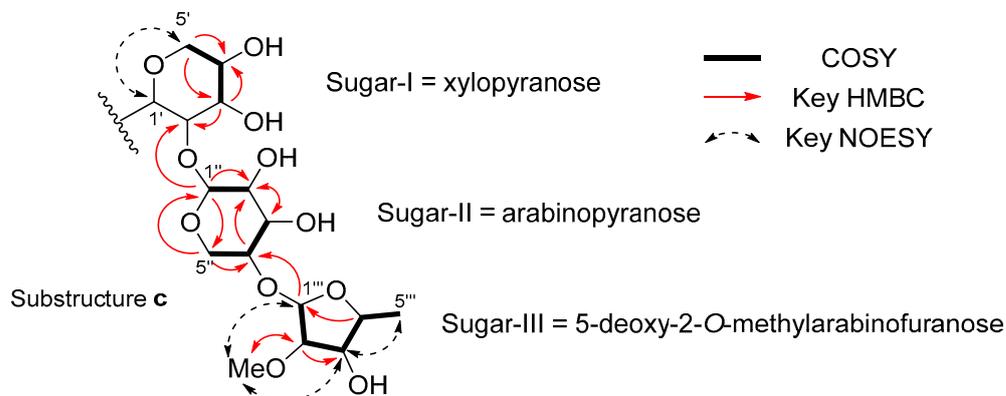


Figure 3. Substructure **c** deduced from COSY and key HMBC and NOESY correlations.

The remaining substructure **b** was composed of $C_7H_6N_2O_3$ and deduced as follows: the NMR spectra in CD_3OD showed a $CHCH_2$ spin system with a broadened H-4 proton (δ_H 4.33, br). The HMBC correlations for H-5a/C-6 (δ_H 2.67, dd, $J = 15.4, 7.6$ Hz/ δ_C 174.5) and primary amide protons (1.4 H integration, δ_H 7.44, s) showing an NOE to H-5a (δ_H 2.60, dd, $J = 14.0, 4.4$ Hz) in CD_3COCD_3 indicated that an amide carbonyl group was connected to C-5. Although four signals for C-1 to C-4 were not observed clearly in the ^{13}C NMR spectrum, the remaining constituents (one proton, five carbons, one nitrogen, and three oxygens) are typical for a tetramic acid moiety with keto-enol tautomerism [21], thus completing partial structure **b** (Figure 4).

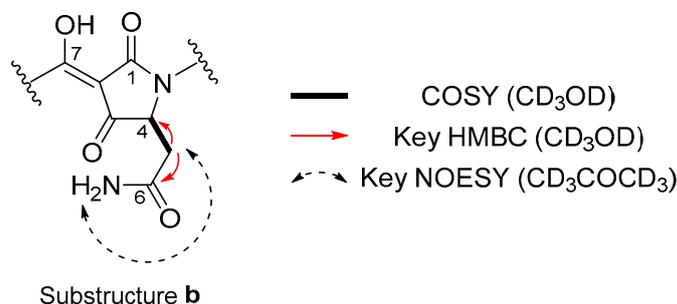


Figure 4. Substructure **b** deduced from COSY and key HMBC and NOESY correlations.

The whole planar structure of **1** was constructed using HMBC and MS/MS data analysis. HMBC correlations for H-8/C-7 (δ_H 7.22, d, $J = 15.1$ Hz/ δ_C 175.2) and H-9/C-7 (δ_H 7.60, dd, $J = 15.1, 11.5$ Hz/ δ_C 175.2) observed in CD_3OD at 320 K indicated that partial structures **a** and **b** were connected between C-8 and C-7. MS/MS (positive ion mode) fragmentation analysis resolved the connection of partial structures **a**, **b**, and **c** and the sequence of the trisaccharide. The ion giving m/z 865.2193 (composed of $C_{38}H_{48}Cl_2N_2O_{15}Na$) was chosen as the precursor ion for the experiment. The intensity of the sodium-cationized ion peak at m/z 599.1586 (calcd for $C_{23}H_{32}N_2O_{15}Na$, 599.1700) was the strongest among the fragment ion peaks observed, suggesting the conjugated system in the tetramic acid moiety was formed by the desorption of substructure **a** and stabilized in MS/MS fragmentation. The second strongest fragment ion peak appeared as m/z 337.0563 (calcd for $C_{12}H_{14}N_2O_8Na$, 337.0648), which was thought to be composed of the tetramic acid moiety (m/z 205.0161) and the xylopyranose (sugar-I). Besides these peaks, fragment ion peaks at m/z 603.1158 (intermediate ion peaks from m/z 865.2193 to 337.0563) and m/z 417.1274 to 285.0872 corresponding to the sugar sequence also supported the structure deduced by NMR experiments (see Figure 5). These experiments confirmed the planar structure of compound **1** as a new tetramic acid glycoside, aurantoside L (**1**) (Figure 5).

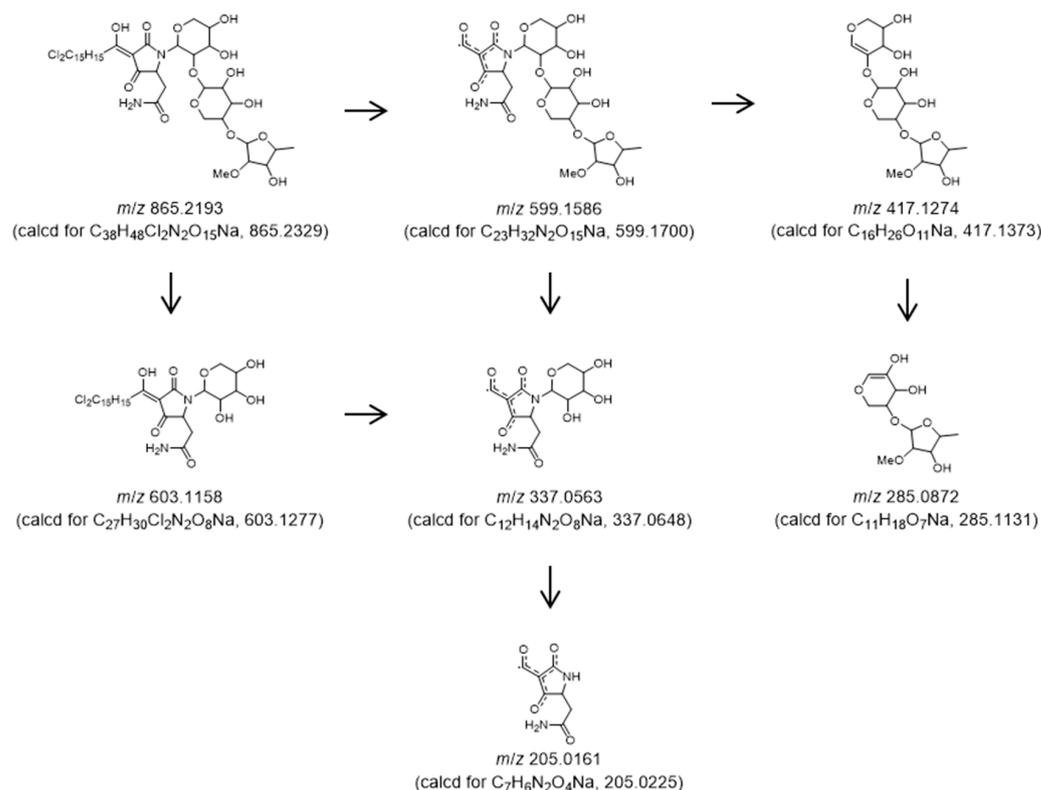


Figure 5. MS/MS fragmentations of aurantoside L (**1**).

According to the literature, aurantosides A-F and rubrosides A-H are all derived from L-aspartic acid (detected via the GC analysis of the acid hydrolysate of the Lemieux oxidation product) and carry D-saccharides [12,13,21–25]. This structural information indicates that a common biosynthetic pathway produces these polyene tetramic acid glycosides. Based on biogenetic reasoning and comparing spectroscopic data with those of analogs, the absolute configurations at C-4 and each saccharide were presumed to be identical to that of analogs. Therefore, the absolute configuration of aurantoside L (**1**) was tentatively assigned as 4S and the saccharides as D-forms.

Aurantoside L (**1**) exhibited anti-leishmanial activity against *La/egfp* with an IC_{50} value of 0.74 μ M, while it showed modest cytotoxicity against HeLa cells and P388 cells with IC_{50} values of 2.4 and 1.1 μ M, respectively. In contrast, aurantoside L (**1**) was inactive at 3.0 μ M against *Trypanosoma congolense*, indicating selective anti-parasitic activity within the same family of Trypanosomatidae.

3. Materials and Methods

3.1. General Experimental Procedures

NMR spectra were recorded on an Avance (400 MHz) spectrometer (Bruker Corporation, Billerica, MA, USA). 1H and ^{13}C NMR chemical shifts were referenced to the CD_3OD solvent peaks δ_H 3.31 and δ_C 49.15 (Wako, Osaka, Japan). HRESI-MS spectra were measured on an Exactive Plus (Thermo Fisher Scientific Inc., Waltham, MA, USA). ESIMS/MS spectra were measured on a TripleTOF 4600 (AB Sciex Pte. Ltd., Tokyo, Japan) in the positive mode. Optical rotation was determined on a P-2200 polarimeter (JASCO Corporation, Tokyo, Japan) in CH_3OH . UV spectra were recorded using a V-630 spectrophotometer (JASCO). IR spectra were measured on a Nicolet6700 spectrometer (Thermo Fisher Scientific Inc.).

3.2. Biological Material

S. japonica was collected by hand using SCUBA (13 m depth, on rocky shores), Tsushima Is., Nagasaki Prefecture, Japan (N 34°15'30", E 129°19'50") in June 2008. The sample was immediately frozen and stored at -25 °C until extraction.

3.3. Isolation

The frozen *S. japonica* (170 g, wet weight) was extracted with CH₃OH and then with CHCl₃/CH₃OH (1:1). The concentrated extract was suspended in H₂O and extracted with CHCl₃ and *n*-C₄H₉OH. The CHCl₃ and *n*-C₄H₉OH layers were combined and subjected to the Kupchan procedure to yield *n*-hexane, CHCl₃, and aqueous CH₃OH layers. The aqueous CH₃OH layer was concentrated to dryness and separated by ODS flash chromatography (H₂O/CH₃OH = 100/0, 80/20, 50/50, 30/70, and 0/100, and CHCl₃/CH₃OH/H₂O = 60/40/10) to give six fractions (fr. A–F). Among them, active fr. E was purified with ODS HPLC (COSMOSIL 5C₁₈ AR-II 20 × 250 mm, 50% CH₃CN + 0.05% TFA, 8 mL/min) to afford 12.5 mg of aurantioside L (**1**, 7.4 × 10^{−3}% yield based on the wet weight).

Aurantioside L (**1**): red amorphous solid; [α]_D²³ −36° (c 0.001, CH₃OH); UV (H₂O) λ_{max} (log ε) 427 (3.79), 250 (3.56) nm; UV (0.01 N HCl) λ_{max} (log ε) 428 (3.77), 329 (3.65) nm; UV (0.01 N NaOH) λ_{max} (log ε) 435 (4.90), 251 (4.24) nm; IR (KBr film) ν_{max} 3350, 1613, 1576, 1530, 1404, 1073, and 1005 cm^{−1}; HRESIMS *m/z* 865.2310 [M + Na]⁺ (calcd for C₃₈H₄₈³⁵Cl₂N₂O₁₅Na, 865.2324. Δ −1.6 mmu); ¹H and ¹³C NMR data; see Table 1.

3.4. Anti-Leishmanial Assay

La/egfp promastigotes (1 × 10⁵ cells) were cultured for 72 h in 199 medium (NISSUI Pharmaceutical, Tokyo, Japan) in 96-well plates with various concentrations of marine invertebrate extracts, as previously reported [26]. Fluorescence was measured with excitation at 485 nm and emission at 538 nm.

3.5. Anti-Trypanosomal Assay

The procyclic form of the parasite (2 × 10⁵ cells per well) *Trypanosoma congolense* IL 3000 was cultured for 48 h in TVM-1 medium [27] in 96-well plates with various concentrations of aurantioside L (**1**). Ten microliters of TetraColor ONE (Seikagaku Biobusiness, Tokyo, Japan) was added to each well. After 4 h, the absorbance of the samples was read at 450 nm using a microplate reader.

3.6. Cell Culture

HeLa human cervical cancer cells were cultured at 37 °C under an atmosphere of 5% CO₂ in Dulbecco's modified Eagle's medium (DMEM, Low Glucose, Wako, Osaka, Japan), containing 10% fetal bovine serum (FBS, Biowest, Nuaillé, France), 2 µg/mL of gentamicin, and 10 µg/mL of antibiotic-antimycotic. P388 murine leukemia cells were propagated and maintained at 37 °C under an atmosphere of 5% CO₂ in Roswell Park Memorial Institute medium (RPMI, Wako), containing HRDS solution (2,2'-dithiobisethanol) and kanamycin sulfate.

3.7. Cytotoxic Test

HeLa human cervical cancer cells in DMEM (cell concentration, 10,000 cells/mL, 200 µL) were added to each well of the microplates (96-well microplates, Costar, Washington, DC, USA) and kept in the incubator at 37 °C under an atmosphere of 5% CO₂ for 24 h. The sample solution (2 µL in MeOH or DMSO) at 1 mg/mL was added to each well with the medium. As the positive control, 2 µL of 1 mg/mL adriamycin was added to a well of each microplate. One-fourth of this medium (ca. 50 µL) with a sample was transferred to a second well with medium (200 µL) to give a 1/5 dilution of the sample concentration. Two or six additional dilution steps gave four or eight sample concentrations. The prepared sample solutions (200 µL) were transferred to wells seeded with HeLa cells and then cultured at 37 °C under an atmosphere of CO₂ for 72 h. Cytotoxic tests against P388 murine leukemia cells were carried out in the same manner except for the medium (RPMI medium, as described in Cell Culture). After 72 h of cultivation, 50 µL of 3-(4,5-dimethyl-2-thiazoyl)-2,5-diphenyl-2H tetrazolium bromide (MTT) saline solution (1 mg/mL) was added to each well and the sample further incubated at 37 °C under an atmosphere of 5% CO₂. After 4 h, the medium was removed via aspiration, and 150 µL

of CH₃COCH₃ was added to each well to lyse the cells. The concentration of the reduced MTT was quantified by measuring the absorbance at 650 nm to estimate IC₅₀ values.

Table 1. NMR data of aurantoside L (1) in CD₃OD at 297 K.

Position	δ_C	δ_H Mult. (J in Hz)	COSY	HMBC
1				
2				
3				
4		4.33 br	5a, 5b	
5a	38.4	2.67 dd (15.4, 7.6)	4, 5b	6
5b		2.81 dd (15.4, 3.6)	4, 5a	6
6	174.5			
7	175.2			
8	122.3	7.22 d (15.1)	9	7, 10
9	146.5	7.60 dd (15.1, 11.5)	8, 10	7, 11
10	133.6	6.61 dd (14.5, 11.5)	9, 11	12
11	145.3	6.89 dd (14.5, 11.6)	10, 12	9
12	135.9	6.56 m	11	13
13	140.5	6.72 dd (16.1, 11.6)		11, 15
14	137.7	6.59 m		16
15	132.9	6.86 dd (14.2, 11.6)		13, 17
16	131.3	6.55 d (11.6)		14, 17, 18
17	134.7			
18	131.8	6.44 d (14.4)	19	16, 17, 20
19	128.0	6.75 dd (14.4, 11.5)	18, 20	17, 21
20	129.0	6.39 d (11.5)	19, 22	18, 21, 22
21	136.4			
22	21.8	2.26 brs	20	20, 21
1'	86.3	4.52 br		
2'	81.4	4.52 br		
3'	79.4	3.48 dd (8.8, 8.8)	4'	2', 4'
4'	70.6	3.62 m	3', 5'a, 5'b	
5'a	69.3	3.21 dd (11.0, 10.8)	4', 5'b	3', 4'
5'b		3.88 dd (10.8, 4.8)	4', 5'a	4'
1''	104.0	5.04 br	2''	2', 2'', 5''
2''	71.7	3.80 dd (9.5, 2.9)	1''	3''
3''	70.9	3.79 dd (9.5, 2.9)	4''	2''
4''	76.1	3.91 m	3'', 5''a, 5''b	2''
5''a	61.6	3.59 dd (12.9, 2.3)	4'', 5''b	1'', 3'', 4''
5''b		3.74 dd (12.9, 0.9)	4'', 5''a	
1'''	98.9	5.09 d (4.4)	2'''	4'', 2''', 3'''
2'''	87.4	3.67 dd (8.1, 4.4)	1''', 3'''	1''', 3''', OMe
3'''	79.9	3.90 dd (8.1, 7.1)	2''', 4'''	1''', 2''', 4''', 5'''
4'''	79.7	3.76 qd (7.1, 6.4)	3''', 5'''	1''', 3'''
5'''	21.0	1.32 d (6.4)	4'''	4'''
OCH ₃	58.5	3.35 s		2'''

4. Conclusions

Bioassay-guided isolation for anti-leishmanial activity afforded a new tetramic acid glycoside, aurantoside L (1), from the marine sponge *S. japonica*. The structure was elucidated using NMR and MS analyses. By combining high-temperature measurements and 2D NMR in different deuterated NMR solvents, the broadened ¹H signals and the unobserved ¹³C signals that are measured by 1D NMR in CD₃OD at room temperature were successfully assigned. Since the MS/MS experiment gives remarkably characteristic fragment ions for tetramic acid glycosides with polyene side chains, it was found to be useful for the structural analysis of compounds containing a tetramic acid moiety such as aurantoside analogs. Cytotoxicity against leukemia cells and antifungal activity have been reported for aurantoside analogs so far. There is a report that the number and structure of saccharide moiety are related to the strength and selectivity of bioactivity [21]. Notably,

this is the first report of a tetramic acid glycoside exhibiting anti-leishmanial activity [28]. The unique structure and strong activity of aurantioside L (1) indicate a novel mechanism of action, which may lead to the development of a new treatment of leishmaniasis.

Supplementary Materials: The following supporting information can be downloaded at <https://www.mdpi.com/article/10.3390/md22040171/s1>: Figures S1–S19: NMR and MS spectra of compound 1.

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Sample Availability: Samples of the compounds are available from the authors.

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